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(54) Title: PREPARATION OF CONCRETE ACCELERATOR

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Preparation of concrete accelerator

This invention relates to cementitious compositions and to accelerators for use therein, particularly for sprayed concrete.

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The application of concrete to a substrate by spraying from a nozzle (commonly referred to as "shotcreting") is a well-established technology, and is widely used in such applications as the lining of tunnels. It is important that the sprayed concrete set very rapidly on the substrate, and this is achieved by the addition to the concrete at the nozzle of an accelerator. These accelerators are quite different from those used with conventional concrete and have traditionally included such materials as alkali metal hydroxides, aluminates and silicates.

The highly alkaline nature of these materials has given handing problems. It also means that their use in confined spaces such as tunnels has led to very unpleasant working atmospheres. Recent attempts to avoid such materials have involved the use of aluminium compounds and typical examples may be found in European Patents 0 076 927, 0 775 097 and 0 742 179, Australian Patent 706917 and European Applications 0 812 812 and 0 946 451.

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It has now been found that it is possible to prepare an accelerator for sprayed concrete by a simple process, which accelerator performs especially well. The invention therefore provides a method of preparing an accelerator for sprayed concrete consisting essentially of the steps of

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- dissolving aluminium sulphate and aluminium hydroxide in water which optionally contains at least one amine dissolved therein, to give a clear solution; and
- optionally adding at least one of at least one stabiliser and at least one defoaming agent;

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the proportions of ingredients present being such that the final product contains from 3%-12% by weight of aluminium sulphate (measured as Al₂O₃), up to 30% by weight of amorphous aluminium hydroxide, up to 15% by weight amine, up to 3% by weight defoaming agent and up to 0.06 mol/kg. stabiliser, the stabiliser being selected from hydroxycarboxylic acids, phosphoric acids and non-alkaline salts of phosphoric acids.

The invention additionally provides an accelerator for use with sprayed concrete prepared by such a method.

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The aluminium sulphate used may be any commercially-available material.

Aluminium sulphates differ in their purity and constitution, the most common being so-called "17%" because it contains 17% of Al₂O₃. In practical terms, the weight percentage of 17% aluminium sulphate, Al₂(SO₄)₃. 14.3 H₂O, which should be used in the process according to the invention lies in the range of from 30% to 60%, preferably from 40%-48%.

The aluminium hydroxide may be any commercially-available amorphous aluminium hydroxide. Although all such aluminium hydroxides will give satisfactory results, it is generally true that the more recent the date of manufacture, the better the result. In addition, aluminium hydroxides which, as a result of their particular manner of manufacture, contain a small proportion of aluminium carbonate (up to 5%) are easier to dissolve and are preferred materials. This behaviour is not obtained by simply adding aluminium carbonate to pure aluminium hydroxide. Although very small quantities of aluminium hydroxide may be used (less than 0.1% is possible), a significant improvement is observed at 5% or more. The preferred range of weight proportions is from 8-25%, preferably from 15-25%.

Although aluminium sulphate, aluminium hydroxide and water can, when utilised together in the process of the invention, give accelerators with good properties, the properties can be considerably enhanced by the use of one or more of three optional, but preferred, components.

The first of these is amine. This must be water-soluble, otherwise there is no restriction on the choice of amine. Preferred amines are alkanolamines, such as diglycolamine, diethanolamine and triethanolamine, diethanolamine being particularly preferred. Up to 10% by weight amine may be used, preferably from 4-7%.

The second preferred additional component is stabiliser, which may be added at the end of the process. This is a material which prevents the aluminium hydroxide/aluminium sulphate solution either from precipitating or from forming a gel. Without stabiliser, the solution will function well as an accelerator, but it will often lack stability and therefore shelf life, necessitating its use very shortly after manufacture, something usually not practical. It is possible and permissible to use more than one stabiliser.

The stabilisers for use in this invention are hydroxycarboxylic acids, phosphoric acids and non-alkaline salts of phosphoric acids. The hydroxycarboxylic acid may be selected from any such acid known to the art. The preferred acid is citric acid, but many other acids, such as lactic acid and ascorbic acid may also be used.

By "phosphoric acid" is meant one of the acids orthophosphoric acid (H₃PO₄),
20 metaphosphoric acid ((HPO₃)_x) and pyrophosphoric acid (H₄P₂O₇). By "non-alkaline
salts" is meant salts which do not include the alkali metals sodium and potassium. Thus,
for example, lithium, calcium and magnesium phosphate salts may be used.

The third preferred additional component, defoaming agent, may be any such material known to the art. Most of these are proprietary commercial materials whose precise composition is never revealed, but any such material known to the art is suitable. Typical examples include silicone types such as AGITAN (trade mark) and fatty acid polyether types such as LUMITEN (trade mark) EL.

The defoaming agent may be used at a rate out up to 5% (solids by weight of the whole composition), preferably from 0.5%-3%. The use of defoaming agent makes the use of less fresh aluminium hydroxides easier. It is believed, without restricting the scope of

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the invention in any way, that its presence helps in the removal of carbon dioxide which accumulates on the surface of the aluminium hydroxide over time. Surprisingly, provided that the defoamer contains no silicone and that it is not present to the extent of more than 3%, it gives an appreciable improvement in setting time over that of an identical composition without defoaming agent or with silicone types.

The process of the invention is readily carried out with standard equipment, and the skilled person will have no difficulty in doing so. It will be appreciated that in order to achieve solutions at the various stages, some heating may be necessary, typically to about 50-60°C.

In the process, the clear solution can be produced by any convenient method. It is possible to add the aluminium sulphate and aluminium hydroxide sequentially in any order to water. It is also possible to add them together to water, or to dissolve or disperse them individually in two different quantities of water and then combine these quantities.

Preferably, the aluminium sulphate and the aluminium hydroxide are added sequentially to water. Preferably the aluminium sulphate is first dissolved in water; aluminium sulphate will dissolve with heating. To this solution the aluminium hydroxide is then added. A clear solution is obtained.

It is possible, although less preferable, first to add the aluminium hydroxide to the water. Aluminium hydroxide does not dissolve readily in water, but gives a fine suspension. To this suspension the aluminium sulphate is added. A clear solution is obtained.

The precise nature of the product of the process is not known. It is certainly not a mere mixture of the original components (the fact that the product is a clear or slightly turbid solution and not an opaque suspension typical of aluminium hydroxide is evidence of this), and without restricting the invention in any way, it is believed to be oligomeric or polymeric in nature.

The accelerator thus prepared gives excellent results when used as a shotcrete accelerator. Shotcrete treated therewith hardens rapidly and has good final strength. The accelerator has a long shelf-life, is resistant to changes in temperature and is completely non-alkaline, thus leading to better working environments.

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The invention is further illustrated by the following non-limiting examples.

Examples 1-3
Preparation of accelerators according to the invention.

The weight proportions used are as follows:

		Example 1	Example 2	Example 3
	17% aluminium sulphate	46%	48%	40%
	amorphous aluminium hydroxide	18%	18%	18%
15	water	30%	28%	28%
	diethanolamine (90% solution)	6%	6%	4%

The diethanolamine is dissolved in the water and the aluminium sulphate is then dissolved in this solution. This is achieved by heating the solution to 50°-60°C and adding with stirring, stirring being continued until a clear solution is obtained. To this heated, stirred solution is gradually added the aluminium hydroxide, and stirring is continued until a clear solution is obtained.

Example 4

An accelerator is prepared by the method and using the materials of Examples 1-3, except that the water content is lowered to 28.7% and there is added 1.3% citric acid monohydrate. This is added after the addition of the aluminium hydroxide, the solution being cooled to room temperature prior to addition. The result is a clear solution.

30 Example 5

Testing of accelerators in mortar.

The mortar used for the testing has the following formulation:

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	Normo 4 Portland cement	450 parts
	SIA 215-1 standard sand	1350 "
	phosphonic acid-based cement	
5	hydration stabiliser ¹	0.3% by weight of cement
	polycarboxylate superplasticiser ²	0.6% by weight of cement

- 1. DELVO (registered trade mark) stabiliser ex MBT
- 2. GLENIUM (registered trade mark) 51 ex MBT

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Sufficient water is added to give a water/cement (w/c) ratio of 0.47.

To samples of the mortar, each of the accelerators of Examples 1, 2 and 4 is added at a rate of 7% by weight of cement, and the initial and final setting times are measured by the Vicat test procedure of EN 196-3. In addition, a commercially-available alkali-free accelerator MEYCO (trade mark) SA 160 was also tested. The results are as follows.

20		Example 1	Example 2	Example 4	Commercial accelerator
	Initial (min)	4	1	3.5	3
	Final (min)	8	5.5	8.5	6

Claims

1. A method of preparing an accelerator for sprayed concrete, consisting essentially of the steps of

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- (i) dissolving aluminium sulphate and aluminium hydroxide in water, optionally containing at least one amine dissolved therein, to give a clear solution; and
- (ii) optionally adding at least one of at least one stabiliser and at least one
 defoaming agent;

the proportions of ingredients present being such that the final product contains from 3%-12% by weight of aluminium sulphate (measured as Al₂O₃), up to 30% by weight of amorphous aluminium hydroxide, up to 15% by weight amine, up to 3% by weight defoaming agent and up to 0.06 mol/kg. stabiliser, the stabiliser being selected from hydroxycarboxylic acids, phosphoric acids and non-alkaline salts of phosphoric acids.

- A method according to claim 1, wherein there is additionally present in the water
 at least one water-soluble amine.
 - 3. A method according to claim 1 or claim 2, wherein there is added at least one of an at least one stabiliser and at least one defoaming agent.
- 25 4. A method according to claim 3, wherein there is added at least one stabiliser.
 - 5. A method according to claim 3, wherein there is added at least one defoaming agent.
- A method according to claim 5, wherein the defoaming agent contains no silicone and is present to a maximum extent of 3%.

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- 7. A method according to any one of claims 1-6, wherein the stabiliser is at least one hydroxycarboxylic acid.
- 8. A method according to claim 1, wherein the clear solution is prepared by the steps of
 - (i) dissolving aluminium sulphate in water, optionally containing at least one amine dissolved therein; and
- 10 (iii) dissolving amorphous aluminium hydroxide in the solution of (i) until a clear solution is obtained.
- A method according to claim 8, wherein there is added to the clear solution at least one of (a) at least one defoaming agent, and (b) at least one stabiliser selected from hydroxycarboxylic acids, phosphoric acids and non-alkaline salts of phosphoric acids, preferably from hydroxycarboxylic acids.
 - 10. An accelerator for use with sprayed concrete, prepared by a process according to any one of claims 1-9.

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- (72) Inventor; and
- (75) Inventor/Applicant (for US only): HOFMANN, Thomas [CH/CH]; Dorfstrasse 22, CH-8166 Niederweningen (CH).
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INTERNATIONAL SEARCH REPORT

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A CLASSI	ミウィナルつか つと くけき ほん	//C04B103/10
A. CLASSI	FIGHTION OF SUBSEC	i may i,cii ,
T 0 0 7	COADOO / 1 A	//COAD102/10
IPC 7	[()4K///14	//1.046103/10

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B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols) IPC 7 C04B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

CHEM ABS Data, EPO-Internal

Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Ε	WO 00 78688 A (HOFMANN THOMAS ;MBT HOLDING AG (CH)) 28 December 2000 (2000-12-28) claims	1,2,4,7, 9,10
Α	WO 96 05150 A (SANDOZ LTD ;SANDOZ AG (DE); SANDOZ AG (AT); LEIKAUF BERNHARD (CH)) 22 February 1996 (1996-02-22) & EP 0 775 097 A 28 May 1997 (1997-05-28) cited in the application	1,2,4
A	EP 0 812 812 A (MBT HOLDING AG) 17 December 1997 (1997-12-17) cited in the application claims; examples/	1,2,4,10

X Further documents are listed in the continuation of box C.	Patent family members are listed in annex.
 Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filling date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filling date but later than the priority date claimed 	 'T' later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention 'X' document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone 'Y' document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art. '3' document member of the same patent family
Date of the actual completion of the international search	Date of mailing of the international search report
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Name and mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk	Authorized officer

INTERNATIONAL SEARCH REPORT

International Application No
PCT/EP 00/12216

		PC1/EP 00/12210
C (Continue	ation) DOCUMENTS CONSIDERED TO BE RELEVANT	
Category °	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 742 179 A (SIKA AG) 13 November 1996 (1996-11-13) cited in the application page 2, line 39 - line 54	1,8
A	DE 196 25 853 A (SANDOZ AG) 2 January 1997 (1997-01-02) claims	1,2,4,7, 9,10
A	claims WO 95 25702 A (RHONE POULENC CHIMIE; BOUYGUES SA (FR); BERTONCINI ANDRE (CH); DUG) 28 September 1995 (1995-09-28) claims	1,5

INTERNATIONAL SEARCH REPORT

information on patent family members

interrumonal Application No PCT/EP 00/12216

Patent document cited in search report		Publication date	1	Patent family member(s)	Publication date
WO 0078688	Α	28-12-2000	NO	20010774 A	15-02-2001
WO 9605150	Α	22-02-1996	AT	174887 T	15-01-1999
			AU	701619 B	04-02-1999
			AU	3257895 A	07-03-1996
			DE	69506881 D	04-02-1999
			DE	69506881 T	02-06-1999
			EP	0775097 A	28-05-1997
			ES	2125644 T	01-03-1999
			FI	970551 A	07-02-1997
			JP	10504273 T	28-04-1998
			US	5968256 A 	19-10-1999
EP 0812812	Α	17-12-1997	AU	723970 B	07-09-2000
			AU	2369697 A	18-12-1997
			BR	9703581 A	22-09-1998
			CN JP	1171378 A 10087358 A	28-01-1998 07-04-1998
			NO	972740 A	15-12-1997
			US	5935318 A	10-08-1999
EP 0742179	Α	13-11-1996	AT	185540 T	15-10-1999
			DE	59603322 D	18-11-1999
			DK	742179 T	10-04-2000
			ES	2139272 T	01-02-2000
			GR NO	3032188 T 961799 A	27-04-2000
			NO TR	961799 A 960992 A	11-11-1996 21-11-1996
					21-11 1990
DE 19625853	Α	02-01-1997	AT	402920 B	25-09-1997
			AT	113896 A	15-02-1997
			AU	706917 B	01-07-1999
			AU	5617796 A	09-01-1997
			CH	691123 A	30-04-2001
			ES	2121689 A	01-12-1998
			FR	2736046 A	03-01-1997
			GB GB	2302688 A,B 96100216 A	29-01-1997 28-02-1997
			GR IT	RM960447 A	28-02-1997 29-12-1997
			JP	9012350 A	14-01-1997
			US	5997630 A	07-12-1999
W0 9525702	 A	28-09-1995	 FR	 2717416 A	22-09-1995
MO 3323/02	^	70 03-1333	AU	1953795 A	09-10-1995

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